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3a-Tosyloxymethyltropane(N^8 -B)borane

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The title compound [alternative name: 8-methyl-8-azabicyclo[3.2.1]octan-3-ylmethyl *p*-toluenesulfonate(N^8-B)-borane], C₁₆H₂₆BNO₃S, has the tosyloxymethyl substituent in an endo position. The BH₃ group is equatorial and the (Nbonded) methyl group is axial, relative to the six-membered heterocycle. The N–B bond of 1.649 (8) Å is one of the longest known.

Comment

 3α -Tosyloxymethyltropane has been synthesized by Murr *et al.* (1992) by hydrogenation of 3-methylenetropane and tosylation of the resulting 3α -oxymethyltropane. Repeating this process as part of a multi-stage synthetic route to littorinetype alkaloids (Tavasli, 1999), we accidentally obtained 3α oxymethyltropane $(N^8 - B)$ -borane, (II), in the first stage and 3α -tosyloxymethyltropane($N^8 - B$)-borane, (I), in the second.



According to the ¹H and ¹³C NMR data, both (I) and (II) formed as 1:1 mixtures of two isomers. One isomer of (I) was isolated and studied in the present work.

The molecule has an *endo* configuration at C3. The BH₃ substituent at the N atom is equatorial and the methyl one axial with respect to the six-membered heterocycle (the opposite may be the case for the unidentified second isomer). A survey of the April 2000 release of the Cambridge Structural Database (Allen & Kennard, 1993) yielded 24 structures containing an N-B bond between a tertiary amine and a BH₃ group, the bond distance ranging from 1.60 to 1.66 Å and

averaging 1.625 Å. Thus, the N-B bond distance of 1.649 (8) Å in (I) is relatively long, probably due to steric overcrowding.

Experimental

HB(CHMeCHMe₂) was prepared by addition of BF₃·OEt₂ (26.7 mmol) to a solution of NaBH₄ (20 mmol) and 2-methyl-2butene (53.4 mmol) in THF (10 ml). 3-Methylenetropane (4.5 mmol) was then added to the solution and the reaction stirred for 16 h at room temperature. A 30% solution of basic peroxide (NaOH/H₂O₂) in ethanol (8 ml) was added and the reaction stirred for another 4 h at 333 K. The mixture was then diluted with water and ether, filtered and the organics were extracted into ether, concentrated in vacuo and the products purified over silica gel, yielding (II) (70 mg, 10%) as a pale-yellow oil. A 1.6 M solution of n-BuLi (2.8 mmol) in hexane was added to a mixture of 2.6 mmol of (II) and 2 mg of Ph₃CH in THF (10 ml) at 273 K and the reaction was stirred for 15 min. A solution of tosyl chloride (2.6 mmol) in THF (5 ml) was then added and the mixture was stirred for 3.5 h at 273 K, diluted with Et₂O and washed with brine. After concentrating the organics in vacuo, a mixture of isomers of (I) was obtained as a white amorphous solid. One isomer was separated by chromatography on silica gel and recrystallized from petroleum ether (55 mg, 8%).

Crystal data

C16H26BNO3S Cu Ka radiation $M_r = 323.25$ Cell parameters from 25 Orthorhombic, Pna21 reflections $\theta = 20.5 - 25.2^{\circ}$ a = 21.443 (2) Å $\mu = 1.731 \text{ mm}^{-1}$ b = 7.959(1) Å c = 10.224 (2) Å T = 293 (2) KV = 1744.9 (4) Å³ Prism, colourless Z = 4 $0.28 \times 0.22 \times 0.11 \ \mathrm{mm}$ $D_x = 1.230 \text{ Mg m}^{-3}$

Data collection

```
Rigaku AFC-6S four-circle diffract-
                                                 1613 reflections with I > 2\sigma(I)
                                                 R_{\rm int} = 0.020
  ometer
2\theta/\omega scans
                                                 \theta_{\rm max} = 75^\circ
                                                 h = -16 \rightarrow 26
Absorption correction: analytical
  (de Meulenaer & Tompa, 1965)
                                                 k = -6 \rightarrow 9
   T_{\min} = 0.726, T_{\max} = 0.837
                                                 l = -8 \rightarrow 12
2243 measured reflections
                                                 3 standard reflections
2157 independent reflections (see
  below)
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Refinement

Refinement on F^2 $w = 1/[\sigma^2(F_o^2) + (0.0473P)^2]$ $R[F^2 > 2\sigma(F^2)] = 0.043$ + 0.4507P] where $P = (F_o^2 + 2F_c^2)/3$ $wR(F^2) = 0.103$ $(\Delta/\sigma)_{\rm max} = -0.001$ S = 1.091 $\Delta \rho_{\rm max} = 0.21 \text{ e } \text{\AA}^{-3}$ 2157 reflections $\Delta \rho_{\rm min} = -0.17 \ {\rm e} \ {\rm \AA}^{-3}$ 208 parameters Absolute structure: Flack (1983) H-atom parameters constrained Flack parameter = 0.35(3)

Table 1

Selected geometric parameters (Å, °).

S-O2	1.424 (4)	C1-C2	1.522 (6)
S-O3	1.432 (4)	C1-C7	1.523 (7)
S-01	1.573 (3)	C2-C3	1.545 (6)
S-C11	1.745 (5)	C3-C9	1.513 (6)
O1-C9	1.467 (5)	C3-C4	1.541 (6)
N-C8	1.497 (7)	C4-C5	1.520 (6)
N-C1	1.502 (6)	C5-C6	1.523 (7)
N-C5	1.518 (6)	C6-C7	1.536 (7)
N-B	1.649 (8)		

every 147 reflections

intensity decay: 0.8%

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C8-N-C1	113.3 (4)	C8-N-B	106.0 (4)
C8-N-C5	112.6 (4)	C1-N-B	113.1 (4)
C1-N-C5	99.7 (3)	C5-N-B	112.3 (4)

1749 independent reflections and 408 of their Friedel equivalents were measured, of which 1229 and 384, respectively, had $I > 2\sigma(I)$. The crystal was a racemic twin with component contributions of 0.65 (3) and 0.35 (3). H atoms were treated as riding and the BH₃ group was treated as a 'rotating body' with the refined B—H distance converging to 1.19 Å.

Data collection: *MSC/AFC Diffractometer Control Software* (Molecular Structure Corporation, 1991); cell refinement: *MSC/AFC Diffractometer Control Software*; data reduction: *TEXSAN* (Molecular Structure Corporation, 1989); program(s) used to solve structure: *SHELXTL* (Bruker, 1997); program(s) used to refine structure:

SHELXTL; software used to prepare material for publication: *SHELXTL*.

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